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ChemQuest 14 Name:
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Information: Valence
Electrons The electrons
in the highest energy
level are called valence
electrons. Valence

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electrons are the electrons located farthest from the nucleus. Valence electrons are always in the highest energy level.

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electrons in the highest
energy level are called
valence

electrons. Valence
electrons are the
electrons located

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farthest from the nucleus. Valence electrons are always in the highest energy level. The valence electrons are the most important electrons in an atom.

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Structures. Example
\\(\PageIndex{2}\\) 1.

Determine the total number of valence electrons in the molecule or ion. Each hydrogen atom (group 1) has one valence electron, carbon (group 14) has 4 valence electrons, and oxygen (group 16) has 6 valence electrons, for a total of $[(2)(1) + 4 + 6] = 12$ valence electrons. 2.

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Date Hour Information:
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energy level are called
valence
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electrons are the
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Arrangement of
Electrons (Shell Model)
Representing Valence
Electrons with Dots

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Part 1 - CHM 1032**

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Information: Steps for Drawing Lewis Structures for Covalent Compounds Study the two examples in the table of how to write structures for CO₃²⁻ and NH₃. Make sure you understand each of the five steps. CO₃²⁻ NH₃

Step #1: Add up the number of valence electrons that should be included in the Lewis Structure. 4

$$4 + 3(6) + 2 = 24$$

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(carbon has four ...

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Homogeneous Critical
Thinking Questions 16.

What is the difference between a "phase of matter" and a "state of matter"? Define each term as best you can.

The "state" of matter tells us whether the substance is a solid, liquid, or gas. A

"phase" of matter is a section of matter that is the ...

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