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Chapter 11 and 12 Study Guide Molarity Molality moles of solute L of solution moles of solute kg of solvent Percent mass grams of solute x 100 grams of solution Mole Fraction mol solute total moles Density of Solution grams solution mL solution X Solution Formation Electrolytic ions present in solution Hydrated ions. interactions keep the ions from recrystallizing Not everything that dissolves in water is an electrolytic solution. o Sugar dissolves.

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