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104 Chapter 4 • The Structure of the Atom Figure 4.2 In his book A New System of Chemical Philosophy, John Dalton presented his symbols for the elements known at that time and their possible combinations. John Dalton Although the concept of the atom was revived in the eighteenth century, it took another hundred years before significant

Chapter 4: The Structure of the Atom
CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure Matching: A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger ____1. Greek thinker; called nature's basic particle an atom, based on the Greek word "atomos" which means "indivisible". Did not have evidence that atoms existed. ____2.

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1 CP Chemistry Review for Chapter 4 Test: Atomic Structure L.O.4.1.1 Explain how Democritus and John Dalton described the atom L.O.4.2.1 Identify types of subatomic particles L.O.4.2.2 Describe the structure of atoms according to the Rutherford atomic model L.O.4.3.1 Explain what makes elements and isotopes different from each other. L.O.4.3.2 Explain how isotopes of an element differ

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